CHROM. 8872

Note

Gas chromatographic separation of isomers of benoxaprofen using liquid crystals

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Benoxaprofen, 2-(4-chlorophenyl)- α -methyl-5-benoxazoleacetic acid (I), is a novel anti-inflammatory agent¹. A ten-stage synthesis of the compound from the raw materials p-chlorobenzoyl chloride and 1-phenylethanol can result in the presence of four possible isomeric impurities (II-V). Attempts to separate low levels of these compounds from benoxaprofen using conventional gas-liquid chromatography (GLC) have proved extremely difficult. As a result the o-chlorophenyl and m-chlorophenyl isomers have been determined by GLC² after alkaline hydrolysis and subsequent methylation, whereas the 6- and 7-propionic acid impurities have been estimated by nuclear magnetic resonance spectroscopy² using a lanthanide shift reagent. We now report the use of a liquid crystal stationary phase to separate the methyl esters of isomers II-V from benoxaprofen methyl ester.

EXPERIMENTAL

A 6-ft. \times 4-mm-I.D. glass column was packed with 4% N,N-bis-(p-methoxy-benzylidene)- α , α' -bi-p-toluidine, a liquid crystal, on Gas-Chrom Q, 100–120 mesh. The column was conditioned at 280° for 24 h before use. The column oven and detector temperatures of the gas chromatograph (Hewlett-Packard Model 5711A equipped with a flame ionisation detector) were maintained at 280° and 300°, respectively. Nitrogen was used as the carrier gas at a flow-rate of 60 ml/min.

Samples of I were methylated with ethereal diazomethane solution before on-column (25 μ g) injection into the gas chromatograph. Determination of low concentrations of isomers II to V required 40-fold attenuation change.

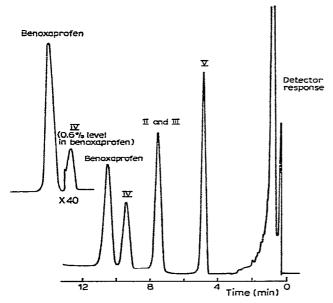


Fig. 1. Gas chromatogram of methyl esters of benoxaprofen and its isomeric impurities on 4% N,N-bis-(p-methoxybenzylidene)- α , α' -bi-p-toluidine.

RESULTS AND DISCUSSION

The liquid crystal N,N-bis-(p-methoxybenzylidene)- α,α' -bi-p-toluidine has a long nematic temperature range³ (181-320°). Maximum selectivity is obtained from the phase at the lower end of its nematic temperature range, although the methyl esters of benoxaprofen and its isomeric impurities then have a long retention time. Consequently a column temperature of 280° was chosen. A 4% phase loading was required to obtain optimum peak symmetry. As the isomeric impurities elute before benoxaprofen, they could be estimated at low levels, as indicated in Fig. 1.

REFERENCES

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